

Phase transition of 3-dimethylammonio-1-phenylpropan-1-one chloride monohydrate

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Key indicators

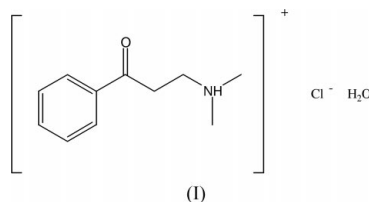
Single-crystal X-ray study
 $T = 173\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$
 R factor = 0.026
 wR factor = 0.066
Data-to-parameter ratio = 18.8For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound, $\text{C}_{11}\text{H}_{16}\text{NO}^+\cdot\text{Cl}^-\cdot\text{H}_2\text{O}$, undergoes a reversible phase transition between room temperature and 173 K. The structure at room temperature crystallizes in space group $Pbca$, with $Z' = 1$, whereas the inversion centre is lost upon cooling and the resulting space group is $P2_12_12_1$, with $Z' = 2$. The low-temperature structure is reported here.

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Comment

The structure of the title compound, (I), has already been determined at room temperature (Schlemper *et al.*, 1982), in space group $Pbca$, with $Z' = 1$. We collected a data set at 173 K and discovered that the reflections that should be absent for the different glide planes in $Pbca$ ($0kl$ for $k = 2n + 1$, $h0l$ for $l = 2n + 1$ and $hk0$ for $h = 2n + 1$) were observed. The only extinction conditions that could be found were $h00$ for $h = 2n + 1$, $0k0$ for $k = 2n + 1$ and $00l$ for $l = 2n + 1$, indicating three twofold screw axes. Therefore, the space group at 173 K must be $P2_12_12_1$, with $Z' = 2$. Refinement of the structure in $P2_12_12_1$ proceeded normally and did not show any signs of missed symmetry; for example, no correlation matrix element was larger than 0.5. A view of the two molecules in the asymmetric unit is shown in Fig. 1.



In order to check whether the phase transition is reversible or not, we collected a data set at room temperature for the previously cooled crystal. The space group was again $Pbca$, showing that the phase transition is reversible. The difference between the two phases is that the inversion centre, which is present at room temperature, is lost upon cooling. However,

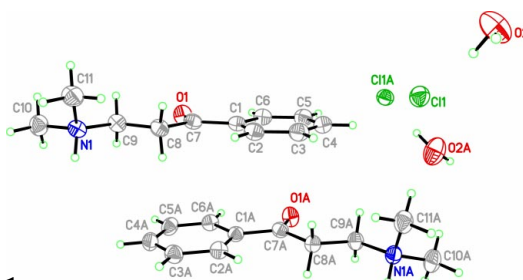


Figure 1

The asymmetric unit of the title compound, with the atom-numbering scheme; displacement ellipsoids are shown at the 50% probability level.

the two molecules in the asymmetric unit are still nearly centrosymmetrically related, as demonstrated by a least-squares fit (Fig. 2). When the phenyl rings are fitted, the side chain of molecule 1 is bent slightly downwards, whereas the side chain of molecule 2 is bent slightly upwards. The structure at room temperature represents an average of the two molecules in the asymmetric unit of (I) (Fig. 3). The Flack (1983) parameter [0.51 (3)] indicates possible inversion twinning.

Experimental

The title compound was synthesized according to the method of Hünig *et al.* (1979).

Crystal data

$C_{11}H_{16}NO^+ \cdot Cl^- \cdot H_2O$	Mo $K\alpha$ radiation
$M_r = 231.71$	Cell parameters from 30714 reflections
Orthorhombic, $P2_12_12_1$	$\theta = 2.1\text{--}27.5^\circ$
$a = 7.3010$ (4) Å	$\mu = 0.30$ mm $^{-1}$
$b = 11.0134$ (7) Å	$T = 173$ (2) K
$c = 29.823$ (2) Å	Plate, colourless
$V = 2398.0$ (3) Å 3	$0.48 \times 0.42 \times 0.24$ mm
$Z = 8$	
$D_x = 1.284$ Mg m $^{-3}$	

Data collection

Stoe IPDS-II two-circle diffractometer	5557 independent reflections
ω scans	4782 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (MULABS; Spek, 1990; Blessing, 1995)	$R_{int} = 0.061$
$T_{min} = 0.869$, $T_{max} = 0.931$	$\theta_{max} = 27.6^\circ$
33 390 measured reflections	$h = -9 \rightarrow 9$
	$k = -14 \rightarrow 14$
	$l = -38 \rightarrow 38$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0406P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.026$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.066$	$(\Delta/\sigma)_{max} < 0.001$
$S = 0.91$	$\Delta\rho_{max} = 0.19$ e Å $^{-3}$
5557 reflections	$\Delta\rho_{min} = -0.11$ e Å $^{-3}$
296 parameters	Absolute structure: Flack (1983),
H atoms treated by a mixture of independent and constrained refinement	2349 Friedel pairs
	Flack parameter = 0.51 (3)

Table 1

Hydrogen-bonding geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$N1-H1 \cdots Cl1A^i$	0.931 (17)	2.189 (17)	3.0778 (13)	159.3 (13)
$O2-H2B \cdots Cl1A^{ii}$	0.84 (3)	2.35 (3)	3.1814 (16)	168 (2)
$N1A-H1A \cdots Cl1^{iii}$	0.92 (2)	2.16 (2)	3.0814 (14)	175.2 (17)
$O2A-H2D \cdots O2^{iv}$	0.90 (3)	1.94 (3)	2.837 (2)	175 (2)
$O2-H2A \cdots Cl1$	0.90 (3)	2.23 (3)	3.1138 (16)	168 (3)
$O2A-H2C \cdots Cl1A$	0.94 (3)	2.32 (3)	3.2353 (16)	163 (2)

Symmetry codes: (i) $1-x, \frac{1}{2}+y, \frac{1}{2}-z$; (ii) $x, 1+y, z$; (iii) $1+x, y, z$; (iv) $\frac{1}{2}+x, \frac{3}{2}-y, -z$.

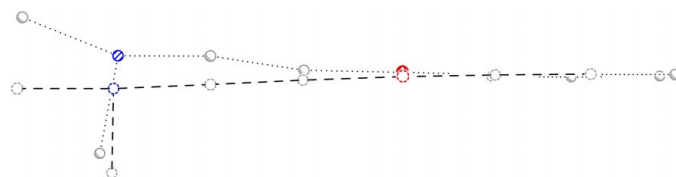


Figure 2

Least-squares fit of the two molecules in the asymmetric unit of (I); only the six C atoms of the phenyl ring were fitted. Cation 1 was inverted prior to fitting. Dashed lines: cation 1; dotted lines: cation 2.

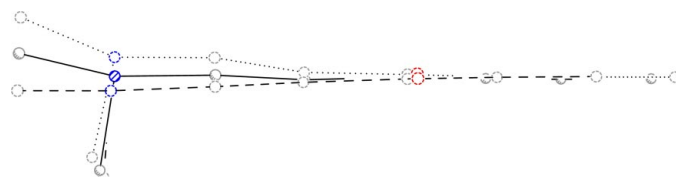


Figure 3

Least-squares fit of the two cations in the asymmetric unit of (I) and the room-temperature structure; only the six C atoms of the phenyl ring were fitted. Cation 1 was inverted prior to fitting. Dashed lines: cation 1; dotted lines: cation 2; solid lines: structure at room temperature.

H atoms bonded to C atoms were included with fixed individual displacement parameters [$U(H) = 1.2 U_{eq}(C)$] using a riding model, with $C_{aromatic}-H = 0.95$ Å, $C_{methylene}-H = 0.99$ Å and $C_{tertiary}-H = 0.98$ Å. H atoms bonded to N and O atoms were refined freely [$N-H = 0.931$ (17) and 0.92 (2) Å, and $O-H = 0.84$ (3)– 0.94 (3) Å].

Data collection: *X-AREA* (Stoe & Cie, 2001); cell refinement: *X-AREA*; data reduction: *X-AREA*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP* in *SHELXTL-Plus* (Sheldrick, 1991); software used to prepare material for publication: *SHELXL97* and *PLATON* (Spek, 2003).

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